

## NOTES

**New Inhibitors of 3 $\alpha$ -Hydroxysteroid  
Dehydrogenase, 0231A and 0231B  
from *Streptomyces* sp. HKI 0231**

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3 $\alpha$ -Hydroxysteroid dehydrogenase is one of the key enzymes involved in inflammatory processes<sup>1)</sup>. Hence, inhibitors of this enzyme are interesting as lead structures for the development of anti-inflammatory agents. In a screening for new microbial metabolites inhibiting 3 $\alpha$ -hydroxysteroid dehydrogenase we disclosed *Streptomyces* sp. HKI 0231 as the producer of new inhibitors, 0231A (**1**)

and 0231B (**2**). Both compounds were formed as minor components in addition to angucyclin-type metabolites such as tetrangulol methylether<sup>2-4)</sup>.

Here we report isolation and structure of these unusual polycyclic metabolites. 10 liters of fermentation broth of *Streptomyces* sp. HKI 0231 were extracted three-times by 10 liters of ethyl acetate. The residue of extract (1.6 g) was subjected to chromatography on Sephadex LH-20 (MeOH) whereby **1** and **2** were eluted together with tetrangulol methylether and other angucycline metabolites. Thereafter this fraction was chromatographed on silica gel 60 (0.063~0.1 mm, column 5 $\times$ 40 cm, elution by CHCl<sub>3</sub>). The tetrangulol methylether was first eluted followed by **1** and **2** as a mixture which appeared as fairly yellowish fractions showing bright fluorescence under UV light ( $\lambda_{\max}$  485~495 nm). Final purification was accomplished by preparative HPLC (Nucleosil RP<sub>18</sub>, 7  $\mu$ m, gradient 95% H<sub>2</sub>O to 95% acetonitrile, 30 minutes) to yield 18 mg of **1** and 15 mg of **2** as yellow solids. The physico-chemical properties of **1** and **2** are shown in Table 1. The UV-VIS

Table 1. Physico-chemical properties of 0231 A (**1**) and 0231 B (**2**).

|  | <b>1</b>   | <b>2</b>  |
|--|--|---|
| Appearance   | yellow solid   | yellow solid  |
| Melting point (uncorr.)                                | 125 °C   | 135 °C  |
| Molecular weight                                       | 361  | 331   |
| HREI-MS  | 361.1296 (M <sup>+</sup> )   | 300.0997 (M-OCH <sub>3</sub> ) <sup>+</sup>   |
| Formula  | C <sub>22</sub> H <sub>19</sub> NO <sub>4</sub><br>(calcd. 361.1278)   | C <sub>21</sub> H <sub>17</sub> NO <sub>3</sub><br>(calcd. 300.1025 for<br>C <sub>20</sub> H <sub>14</sub> NO <sub>2</sub> )                            |
| UV-VIS ( $\lambda_{\max}$ (nm),<br>MeOH)               | 240, 285, 365, 380, 425,<br>440, 462   | 240, 285, 365, 380, 420,<br>440, 460  |
| [ $\alpha$ ] <sub>D</sub> (MeOH; 3.5 mg/ml,<br>0.5 cm) | 0°   | 0°  |
| IR ( $\lambda_{\max}$ , cm <sup>-1</sup> , KBr)        | 744, 795, 852, 885, 978,<br>1026, 1070, 1110, 1134,<br>1184, 1265, 1307, 1334,<br>1402, 1437, 1461, 1484,<br>1555, 1586, 1590, 1627,<br>1720, 2950, 3420 | 737, 790, 837, 867, 897,<br>954, 1025, 1030, 1086,<br>1137, 1190, 1270, 1331,<br>1339, 1399, 1433, 1487,<br>1579, 1585, 1625, 1629,<br>1716, 2986, 3420 |
| Fluorescence ( $\lambda_{\max}$ , in<br>MeOH )         | 493 nm   | 487 nm  |
| ( $\lambda_{\max}$ excitation):                        | (380 nm)   | (440 nm)  |

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and fluorescence spectra attested to the presence of an extended conjugated  $\pi$ -electron system. The absorptions at  $\lambda_{\max}$  1720 and 1716  $\text{cm}^{-1}$  in the IR spectra of **1** and **2** indicated two carbonyl groups in both molecules. Measurement of optical rotation (in MeOH) indicated that both compounds are inactive. Thus compound **2** could be a mixture of enantiomers. The ESI mass spectrum of **1** displayed ion peaks of  $m/z$  362.2 ( $M+H$ )<sup>+</sup>, 745.6 ( $2M+Na$ )<sup>+</sup> and 1105.9 ( $3M+Na$ )<sup>+</sup>. In the ESI-MS of **2** pseudomolecular ions with  $m/z$  332.0 ( $[M+H]^+$ ),  $m/z$  353.7 ( $[M+Na]^+$ ),  $m/z$  685.4 ( $[2M+Na]^+$ ) and  $m/z$  1016.0 ( $[3M+Na]^+$ ) were visible. The molecular weight and the chemical formula of **1** and **2** as shown in Table 1 were furnished by HREI-MS (**1**:  $M^+$ :  $m/z$  361.1296 (80%), calcd. 361.1314 for  $C_{22}H_{19}NO_4$ ; **2**:  $[M-OCH_3]^+$ :  $m/z$  300.09970 (80%), calcd. 300.10248 for  $C_{20}H_{14}NO_2$ ). In addition the HREI-MS of **1** displayed  $m/z$  330.1161 ( $[M-OCH_3]^+$ , 100%, calcd. 330.1130 for  $C_{21}H_{16}NO_3$ ) and  $m/z$  315.0874 ( $[M-OCH_3-CH_3]^+$ , 80%, calcd. 315.0859 for  $C_{20}H_{13}NO_3$ ).

The number, bonding type and multiplicity of carbon atoms in **1** and **2** (Fig. 1) were determined by the  $^{13}\text{C}$  and DEPT NMR spectra. The presence of quaternary carbon signals at 181.6 ppm in **1** and 181.3 ppm in **2** showed the presence of carbonyl groups in conjugation to double bonds. The quaternary carbon signal at 114.2 ppm in the

$^{13}\text{C}$  spectrum of **1** was assigned to an *ortho* carbonic acid amide (C-5). In the  $^{13}\text{C}$  spectrum of **2** a signal at 86.7 ppm attested to a semi aminal-type structure. Moreover, the number of methoxyl and methyl groups in both molecules was determined doubtlessly from the  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra. The positions of protons in **1** and **2** were suggested from the 2D NMR spectra ( $^1\text{H},^1\text{H}$ -COSY, HSQC). In addition to the singlet proton signals of the methyl and methoxyl groups the  $^1\text{H}$  NMR spectrum of **2** displayed 6.77 ppm for H-5. Its unusual deep-field shift can be explained by the electronic effects of the neighbored

Fig. 1. Structures of 0231A (**1**) and 0231B (**2**).

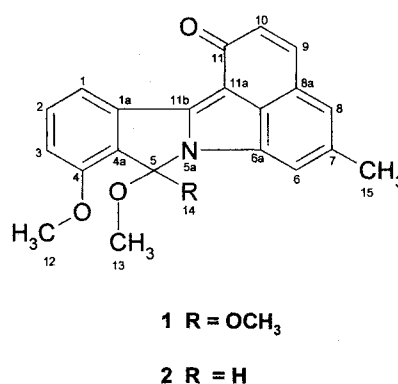
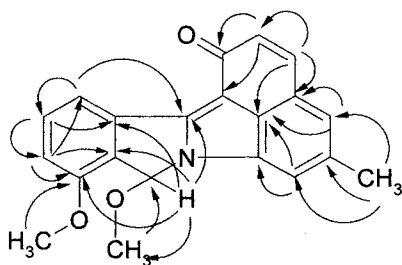


Table 2. Assignment of  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of **1** and **2** (500 MHz,  $\text{CDCl}_3$ , TMS as internal standard,  $\delta$  in ppm,  $J$  in Hz, s: singlet, d: doublet).

| position | <b>1</b>            |                     | <b>2</b>            |                     |
|----------|---------------------|---------------------|---------------------|---------------------|
|          | $\delta_{\text{C}}$ | $\delta_{\text{H}}$ | $\delta_{\text{C}}$ | $\delta_{\text{H}}$ |
| C-1      | 118.2               | 8.22 d, 78.3        | 118.0               | 8.22 d, 7.1         |
| C-2      | 133.8               | 7.61 dd, 8.8, 7.9   | 132.9               | 7.57 dd, 7.1, 8.5   |
| C-3      | 113.6               | 7.06 d, 7.9         | 113.1               | 7.03 d, 8.5         |
| C-4      | 155.7               | -                   | 156.1               | -                   |
| C-4a     | 125.2               | -                   | 127.7               | -                   |
| C-5      | 114.2               | -                   | 86.7                | 6.77 s              |
| C-6a     | 130.3               | -                   | 131.3               | -                   |
| C-6      | 114.2               | 7.54 d, 0.5         | 113.3               | 7.45 d, 0.5         |
| C-7      | 135.3               | -                   | 134.7               | -                   |
| C-8      | 124.0               | 7.35 d, 0.5         | 123.7               | 7.33 d, 0.5         |
| C-8a     | 109.4               | -                   | 109.1               | -                   |
| C-9      | 137.8               | 7.66 d, 9.6         | 137.4               | 7.64 d, 9.6         |
| C-10     | 132.4               | 6.71 d, 9.6         | 132.6               | 6.71 d, 9.6         |
| C-11     | 181.6               | -                   | 181.3               | -                   |
| C-11a    | 124.5               | -                   | 124.5               | -                   |
| C-11b    | 146.1               | -                   | 148.6               | -                   |
| C-12     | 55.9                | 4.00 s              | 55.8                | 3.99 s              |
| C-13     | 52.6                | 3.10 s              | 51.0                | 2.97 s              |
| C-14     | 52.6                | 3.10 s              | -                   | -                   |
| C-15     | 22.1                | 2.58 s              | 22.1                | 2.58 s              |

Fig. 2. Instructive C,H long-range couplings in the HMBC spectrum of **2**.



aromatic ring and the two heteroatoms.

For the assignment of  $^1\text{H}$  and  $^{13}\text{C}$  signals as shown in Table 2 the observable C,H long-range couplings in the HMBC spectra (Fig. 2) were of pivotal importance. The positions of the methoxyl and methyl groups were thus assignable. In the HMBC spectrum of **1** instructive C,H long-range correlations were observed between both H-13 and H-14 with C-5 confirming that this carbon atom belongs to the diester of an *ortho*-carbonic acid amide. The HMBC spectrum of **2** showed C,H long-range couplings of H-5 with C-13, C-4a, C-1a, C-11b and C-4 which proved the position of the semiaminal-type C-5. Correlations between H-5 and C-4 in **2** confirmed the methoxyl substituent at C-12. It can be suggested that its position in **1** is the same. The observable NOE between H-5 (6.77 ppm) and H-6 (7.45 ppm) in the NOESY spectrum supported the structure of **2** as shown in Fig. 1.

Thus the structures of **1** and **2** as new polycyclic heteroaromatic systems were assigned on the basis of mass spectrometry, 1D and 2D NMR measurements (Table 2). The observed bright fluorescence is explainable by the presence of a new conjugated oxonaphthopyrrolo chromophore.

Coproduction of tetrangulol methylether<sup>2-4)</sup> and other angucycline type metabolites suggested that **1** and **2** are shunt metabolites of the angucycline biosynthetic pathway<sup>4)</sup>. Their formation could be explained by the oxidation of a quinone carbonyl, introduction of nitrogen, dehydration and rearrangement of the cyclic system.

Both **1** and **2** displayed no activity against a series of Gram-positive and Gram-negative bacteria and fungi in concentration  $<200\ \mu\text{g/ml}$ . However, 0231A (**1**) and 0231B (**2**) inhibited  $3\alpha$ -hydroxysteroid dehydrogenase with  $\text{IC}_{50} = 10.5\ \mu\text{g/ml}$  and  $2.5\ \mu\text{g/ml}$ , respectively<sup>1)</sup>.

## Experimental

Mass spectra were recorded on a high-resolution sector-field mass spectrometer AMD-402 (AMD Intectra, Harpstedt, Germany), NMR spectra on a Bruker Avance DRX 500 spectrometer, UV-VIS and IR fluorescence spectra on Shimadzu spectrometers. Optical rotation was measured with a Kernchen polarimeter (Dr. KERNCHEN OPTICS, Seelze, Germany).

The strain *Streptomyces* sp. HKI 0231 was isolated from a soil sample collected near the Grotta dei Cervi (Porto Badisco, Italy). It was identified as *Streptomyces* sp. due to its morphological and physiological characteristics and was deposited in the strain collection of the Hans-Knöll-Institute for Natural Products Research Jena (Germany). A seed culture was prepared by inoculating 1~2 cm<sup>2</sup> agar pieces of a surface culture into Erlenmeyer flasks containing 100 ml of a medium composed as follows (g/liter): glucose 15, soybean meal 15, NaCl 5, CaCO<sub>3</sub> 1, KH<sub>2</sub>PO<sub>4</sub> 3, pH 6.5 (28°C, rotary shakers, 180 r.p.m., 48 hours). 5 ml of the seed culture were used to inoculate 100 ml medium in 500 ml Erlenmeyer flasks composed of (g/liter): mannitol 20, soybean meal 20, pH 6.5 (sterilization for 25 minutes at 110°C). Cultivation was carried out for 120 hours on rotary shakers (180 r.p.m., 28°C).

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